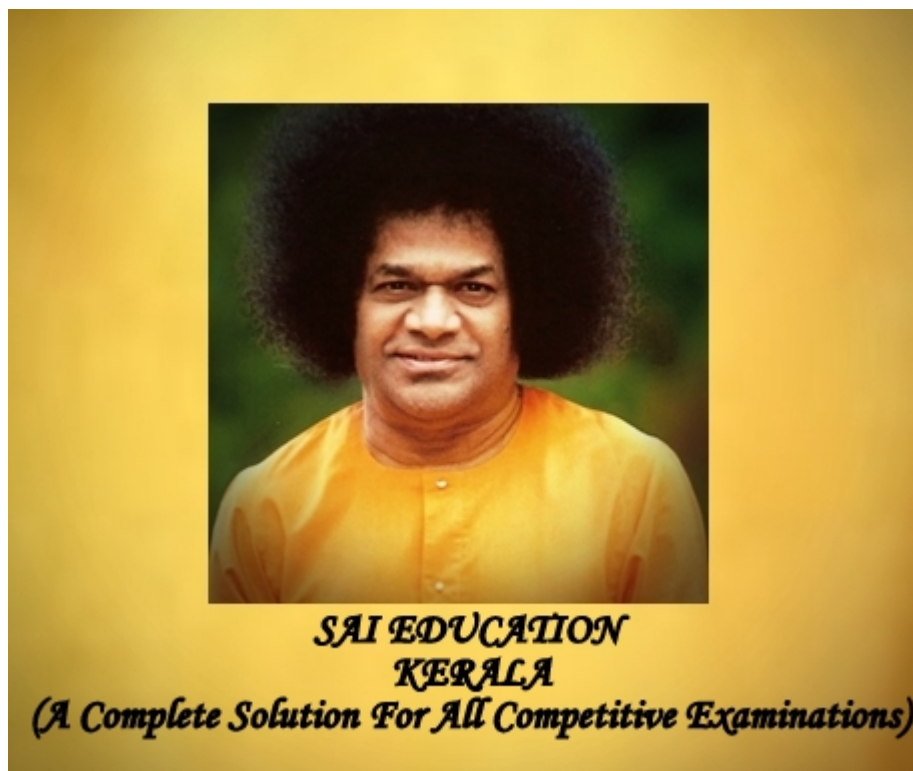


ONLINE COACHING DAY 65 (04-06-19) - CHEMISTRY

KERALA UNIVERSITY ASSISTANT SPECIAL




Name of the Candidate *

M-2 ▼

Place *

Thrissur ▼

Untitled Section

Please watch the online video (1,2 and 3) 

<https://youtu.be/FGXerJiEdxo>

<https://youtu.be/l6boCkw9mJU>

<https://youtu.be/owrqCOTrcf0>

1. Smallest constituent particle in the universe is *

1 point

- Atom
- Molecule
- Compound
- Elements

2. Atoms having same atomic number but different mass number is known as *

1 point

- Isomer
- Isobar
- Isotope
- None of these

3. The total no. of protons or electrons in an atom is known as *

1 point

- Mass number
- Atomic Number
- Isotope
- None of these

4. Value of Avogadro's number *

1 point

- 0.6023×10^{22}
- 6.023×10^{22}
- 6.032×10^{23}
- 6.023×10^{23}

5. Losing of electrons is the chemical property of *

1 point

- Metals
- Non Metals
- Metalloids
- None of these

6. Which law states that energy can neither be created nor be destroyed but can change from one form to another. *

1 point

- Charle's Law
- Law of conservation of Mass
- Law of conservation of Energy
- Faraday's Law

7. Unlike charges _____ each other *

1 point

- repels
- attracts
- comes parallel
- none of these

8. How many electrons should be present in the valence shell to attain stability *

1 point

- 4
- 6
- 8
- 7

9. Which of the following is not sub atomic particles in an atom. *

1 point

- Electrons
- Neutrons
- Quarks
- Protons

10. Flow of electrons from +ve to -ve is known as *

1 point

- Electricity
- Magnetism
- Energy
- Force

11. Ability to do work is known as *

1 point

- Force
- Stress
- Energy
- None of these

12. Isotope of hydrogen which contains one proton and one neutron is *

1 point

- Protium
- Deuterium
- Tritium
- None of these

13. Which of the following has a -ve charge *

1 point

- Protons
- Neutrons
- Electrons
- None of these

14. 17th group elements is known as *

1 point

- Alkali Metals
- Alkali Earth Metals
- Halogens
- Transition Elements

15. Which of the following is the radioactive element in the 1st group elements? *

1 point

- Lithium
- Sodium
- Rubidium
- Francium

16. Neutron was discovered by *

1 point

- James Chadwick
- J. J. Thomson
- Ernest Rutherford
- None of the above

17. Lightest metal *

1 point

- Calcium
- Sodium
- Lithium
- Potassium

18. Which is not a type of elements ? *

1 point

- Metals
- Non Metals
- Metalloids
- Gases

19. What nucleus of atom contains ? *

1 point

- Protons
- Electrons
- Electrons and Protons
- Protons and Neutrons

20. The isotope atoms differ in ? *

1 point

- Number of neutrons
- Atomic number
- Number of electrons
- Atomic weight

21. Mass number is equal to *

1 point

- Number of electrons
- Number of neutron
- Number of protons
- Total number of protons and neutrons

22. Which gas is used in fire extinguishers ? *

1 point

- Carbon dioxide
- Nitrogen oxide
- Carbon monoxide
- Sulpher dioxide

23. The most malleable metal is *

1 point

- Platinum
- Silver
- Gold
- Iron

24. The Chemical Symbol of Sodium is *

1 point

- So
- Sd
- Na
- Nu

25. Which metal is heaviest ? *

1 point

Osmium

Mercury

Iron

Nickle

Thank You!!!



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